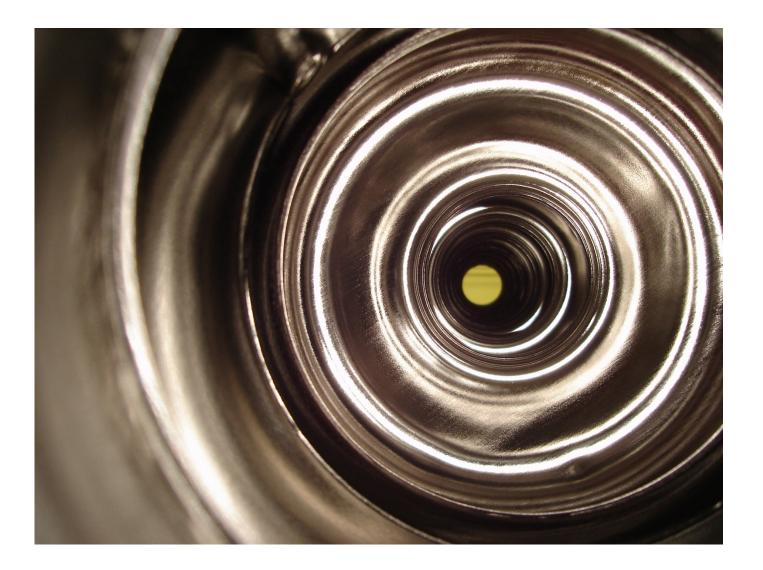
Lecture B2: Superconductive RF

Surface Preparation

T. Saeki (KEK) LC school 2015 27 Oct. - 6 Nov. 2015, Whistler, Canada

Inner Surface Preparation of SC Cavity



Overview of Inner Surface Preparation

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- Thick/Rough Removal (>100 um)
 Three alternative methods:
 Buffered Chemical Polishing (BCP)
 Electro-Polishing (EP)
 Centrifugal Barrel Polishing (CBP)
- Annealing / Degassing (750 800 C, ~3 h)
- Final Thin Removal (10 30 um) Two alternative methods: Buffered Chemical Polishing (BCP) Electro-Polishing (EP)
- High Pressure Rinse (>7 h)
- In-situ Baking (120 140 C, 48 h)

Overview of Inner Surface Preparation

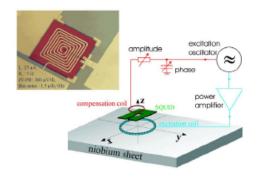
Process Name	Process	Removal Thickness (um)	Purpose
Centrifugal Barrel Polishing (CBP)	Mechanical removal with water and media (stones / plastic)	>100	Removal of damaged layer of Nb, or removal of defects
Buffered Chemical Polishing (BCP)	Chemical Reaction / Etching	Rough: > 100 Final: 10 - 30	Rough: Removal of damaged layer of Nb. Final : Smooth surface
Electro-Polishing (EP)	Electro-Chemical Reaction	Rough: > 100 Final: 10 - 30	Rough: Removal of damaged layer of Nb. Final : Smooth surface
Annealing / Degassing	750 ^o C,3 hours Vacuum Furnace	-	Release of stress in material / Degassing of H
High Pressure Rinse (HPR)	High Pressure Rinse with Ultra Pure Water (UPW)	-	Removal of contamination / Clean surface
In-situ Baking	120 ^o C, 48 hours baking with Vacuum pumping inside of cavity	-	Diffusion of Oxygen

Scanning of defects with eddy current

Successfully developed at DESY to pre-screen Nb Sheets for defects: eddy current, resolution ~ 100 μm

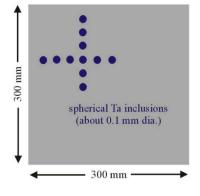


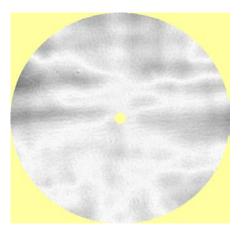
squid, resolution < 50 μ m

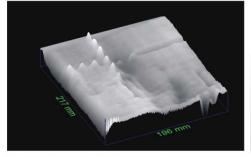


Low Tc superconducting SQUID system for eddy current testing of niobium sheets is in development

(W.Singer, X.Singer)







Two-dimensional distribution of eddy-current field above the niobium test sample, measured from the back side of the sample. The excitation coil had 30 turns and a diamter of 3 mm, the excitation frequency was 10 kHz. The reference phase of the lock-in amplifier was chosen such that the lift-off effect was minimized.

Scanning of defects with eddy current

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Eddy Current Scanning system for SNS high purity niobium scanning



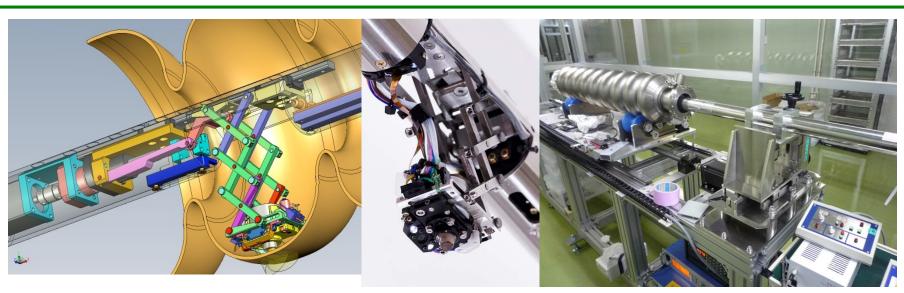




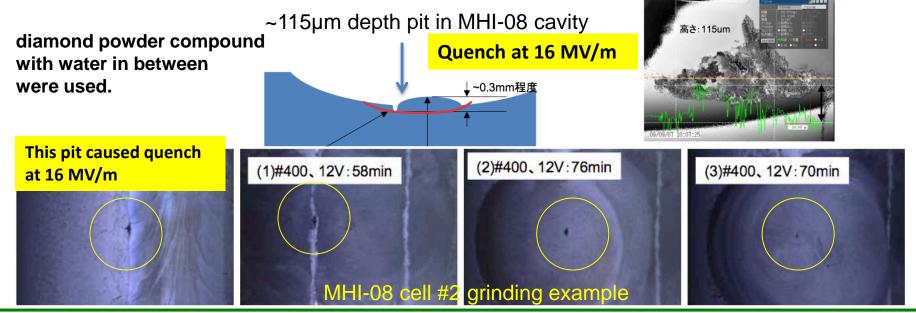




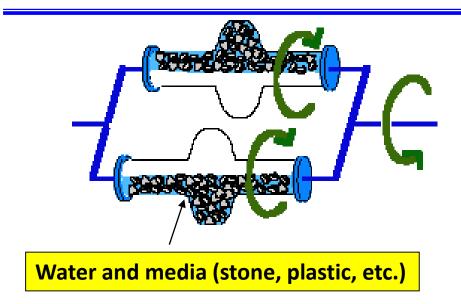
Local grinding at KEK



Grinding only for pit, without touching other surface



Centrifugal Barrel Polishing (CBP) Mechanical grinding







Media : stones (rough removal)

Media : plastic (final removal)

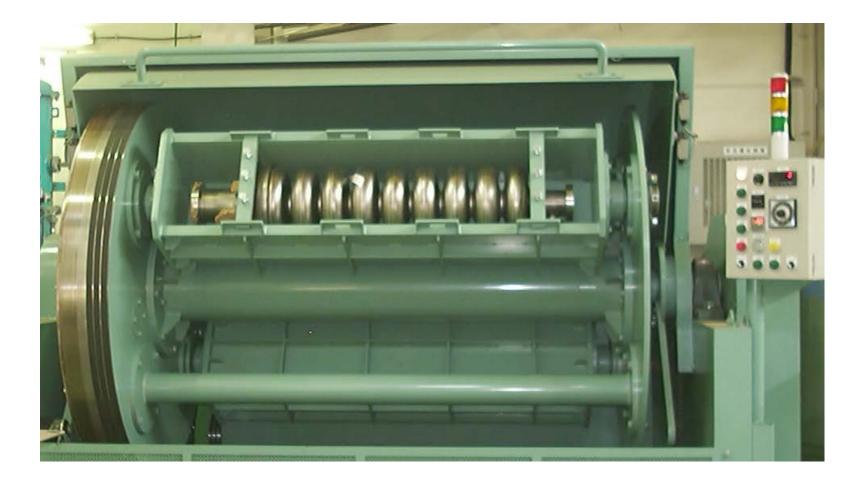
Example (KEK recipe)

Step 1) Rough removal with stones + water : 4 hours x 3 times Removal thickness = 25 um x 3 = 75 um Step 2) Final removal with plastic + water : 4 hours x 3 times Removal thickness = 20 um x 3 = 60 um

Total removal thickness = 135 um

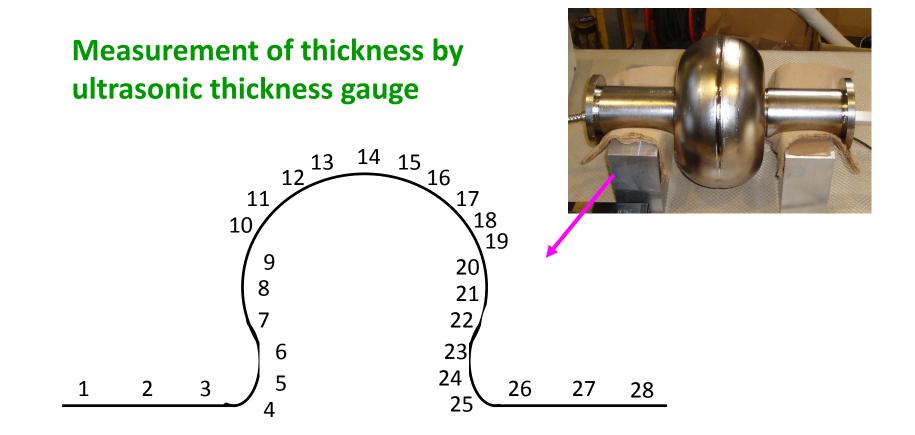


Centrifugal Barrel Polishing (CBP) Mechanical grinding



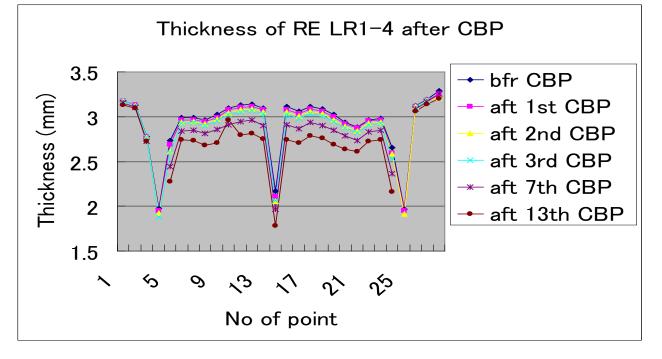


Measurement of Removal Thickness (CBP)

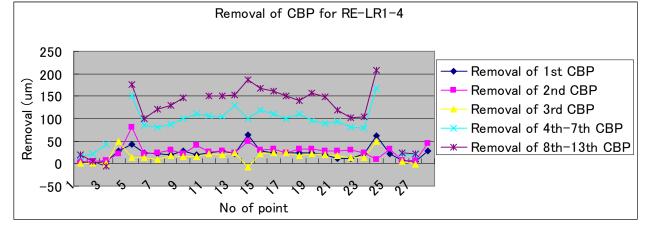


Positions of measurement

Measurement of Removal Thickness (CBP)

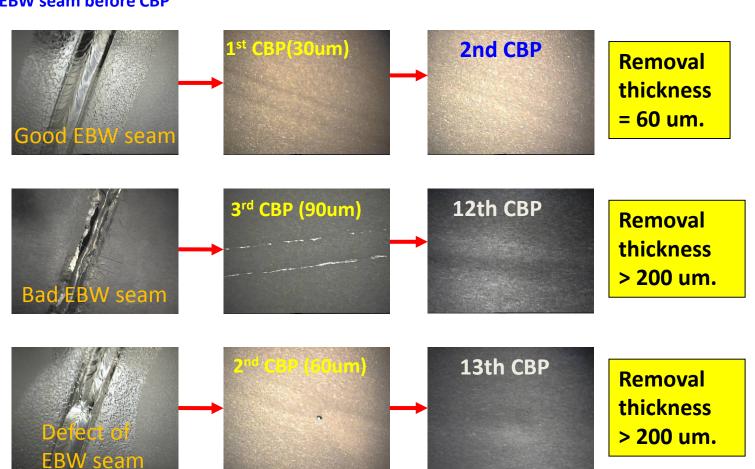


Thickness measurements by ultrasonic thickness gauge



Removed weight 1st CBP: 21.0 g 2nd CBP: 21.0 g 3rd CBP: 16.5 g 4th-7th CBP: 78.0 g 8th-13th CBP: 94.3 g Total of removed weight = 230.8 g

Centrifugal Barrel Polishing (CBP) Mechanical grinding

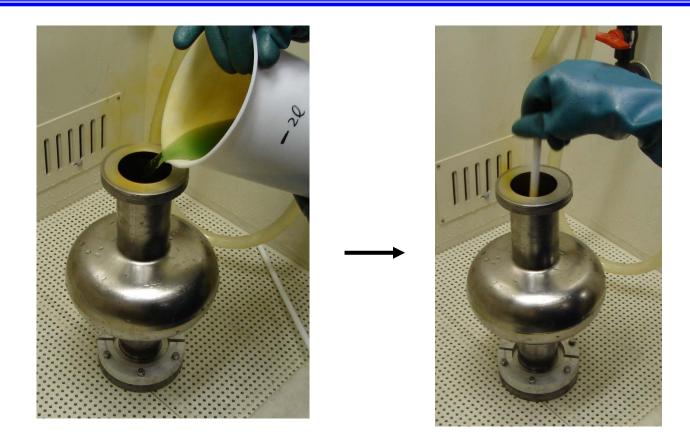


EBW seam before CBP

 \triangleleft

Buffered Chemical Polishing (BCP)



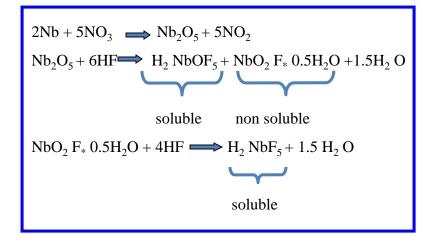


Components of BCP acid (KEK recipe) HF (HF 46%) : HNO_3 (HNO₃ 60%) : H_3PO_4 (H_3PO_4 80%) = 1 : 1 : 1 (Volume Ratio) H_3PO_4 can be increased if you like slow etching (1:1:2, 1:1:3, etc...)

Various BCP systems in the world 🧹

BCP:

```
Mixture of HF/HNO<sub>3</sub>/H<sub>3</sub>PO<sub>4</sub> in ratios 1:1:1 or 1:1:2 @ 10-15C
```



Exothermic reaction Removal rate ~ 2 µm/min @ 10C

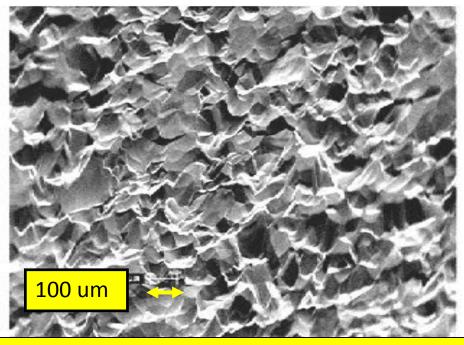






Buffered Chemical Polishing (BCP)





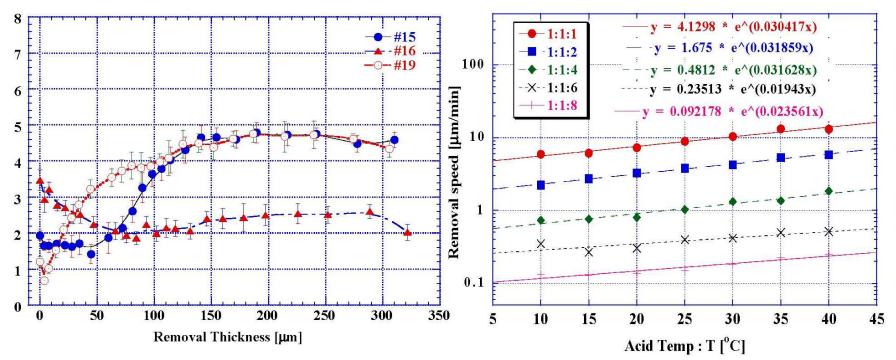
Simpler than EP, but the surface is rougher than EP. The roughness is coming from the difference of etching among the grain. Steps are made along the grain-boundary.

> Chemical reaction: $6Nb+10HNO_3 \Rightarrow 3Nb_2O_5+10NO \ddagger+5H_2O$ $Nb_2O_5+10HF \Rightarrow 2NbF_5+5H_2O$

* 6Nb+10HNO₃+30HF \rightarrow 6NbF₅+10NO[†]+20H₂O

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Buffered Chemical Polishing (BCP)



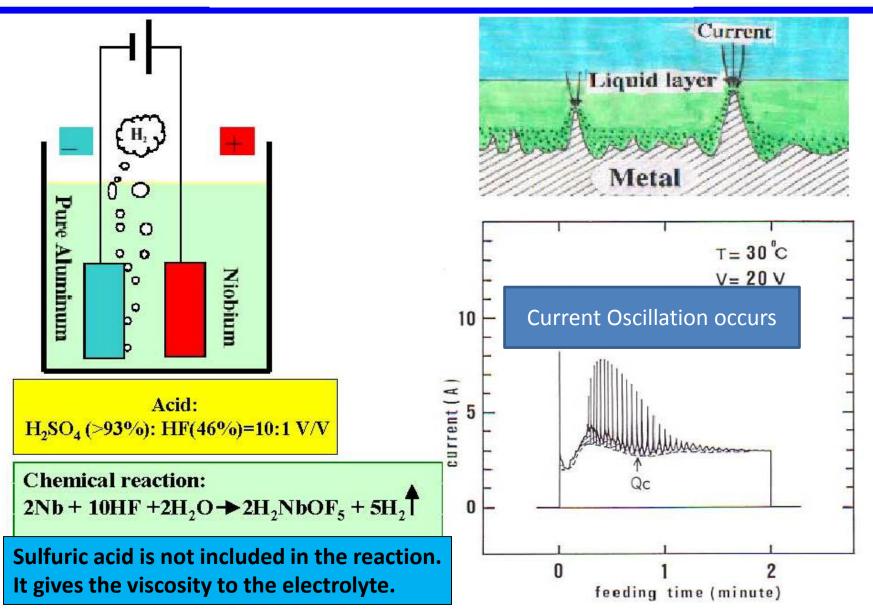
Typical surface roughness = $2 \sim 5 \,\mu m$ after 100 μm CP,

Material removal speed ~ 10μ m/min at the room temperature with CP acid 1:1:1

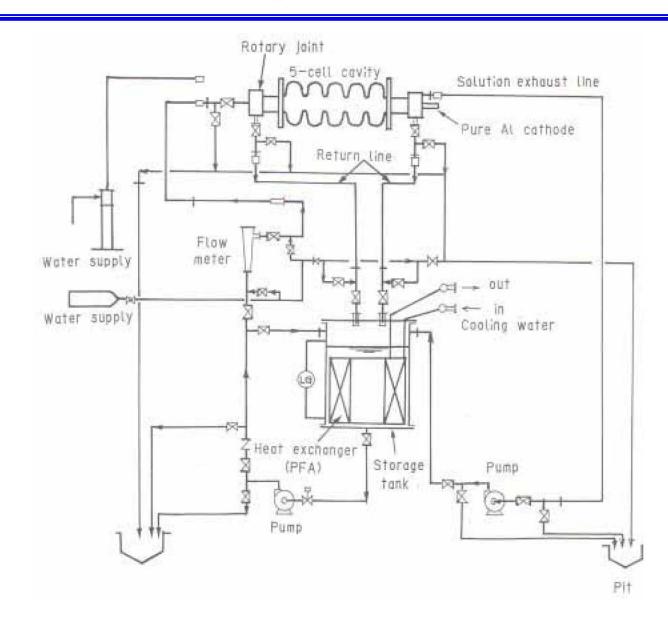
CP is faster in material removing than **EP**.

The roughness is related to the steps along the grain-boundary. And so the toughness changes if the grain-size of Nb material changes.

Electro-Polishing (EP)



EP System Flow

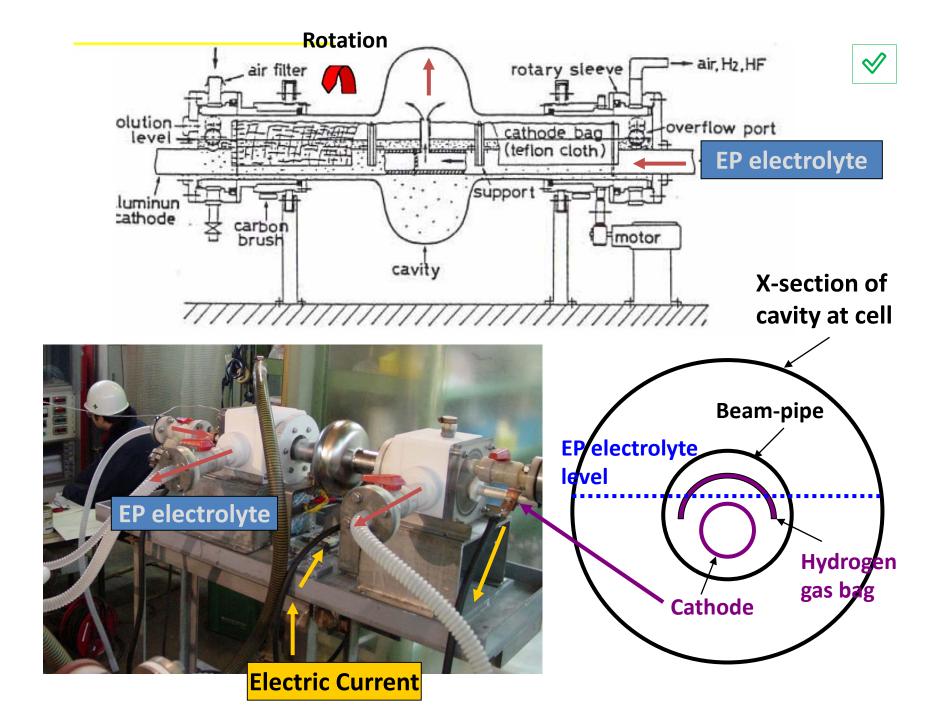


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Horizontal EP system for single-cell cavity

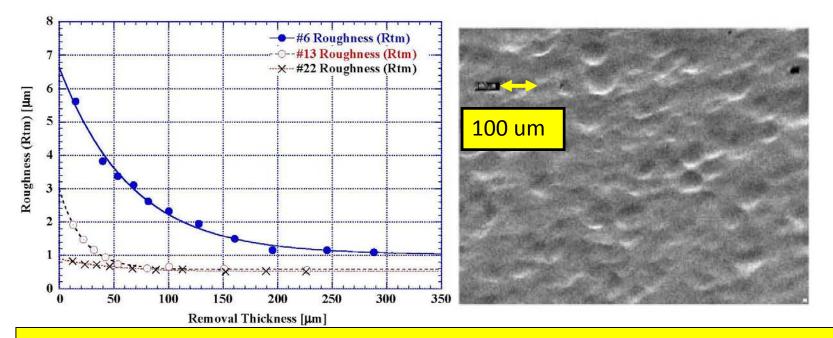
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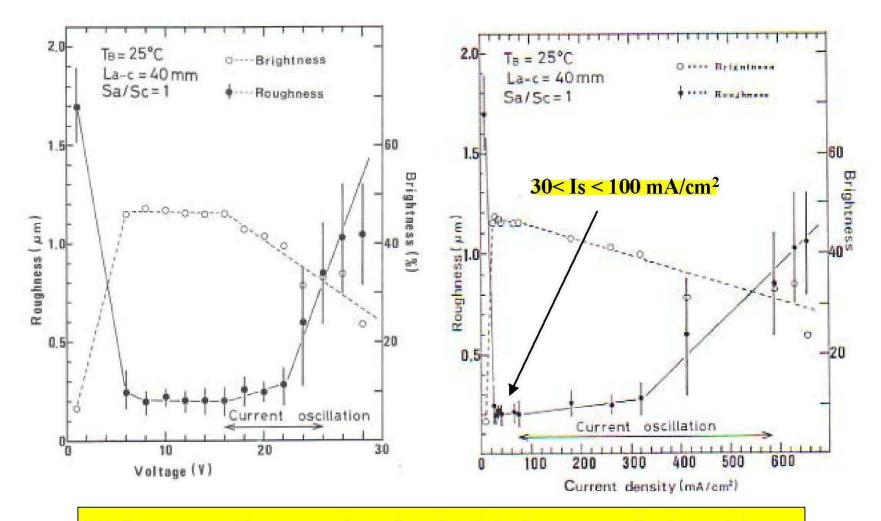


EP Finishing



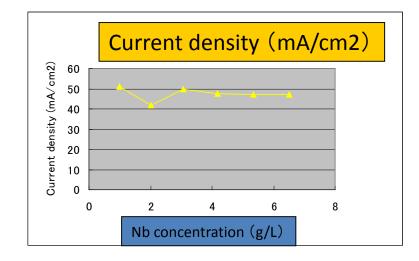
- 1) The final roughness depends on the initial roughness.
- 2) The roughness goes down as the exponential function to the removal.
- 3) Steps are not created along the grain-boundary. This causes smooth surface.
- 4) If voltage is switched off, the process stops. The control of process is easier than BCP.

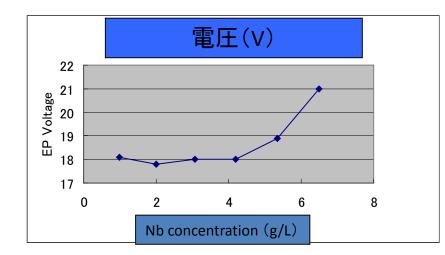
What is the curtail EP parameter?



Voltage or Current density? This is a coupled problem.

Nb concentration in EP electrolyte



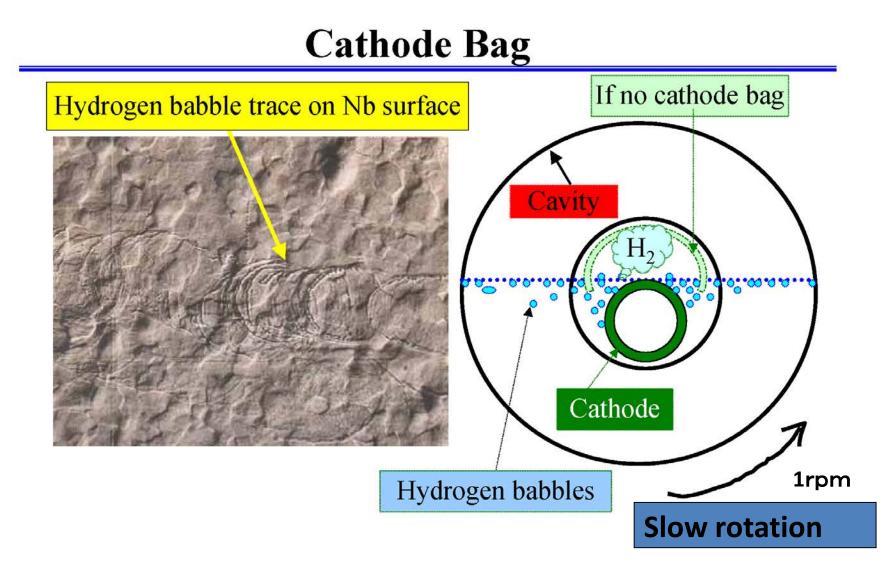


Repeating EP processes under the constant current density(= 50mA/cm²).

If Nb concentration increased beyond 9 g/L, the surface finish is going to be rough. And sulfur contamination increase also.

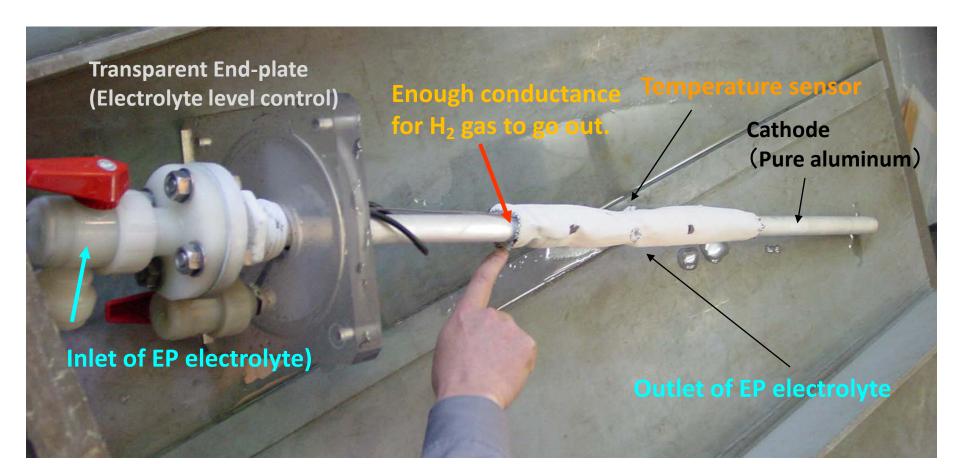
= > If this difficulty is overcome, the cost of mass-production can be lowered.

In this experiment, the voltage increased with the increase of Nb concentration.



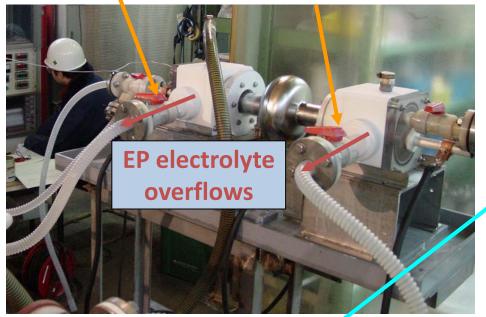
Hydrogen-gas bag should be set on the cathode appropriately. Otherwise, hydrogen-gas goes into the inner wall of cavity and causes Q-disease (low Q from low gradient).

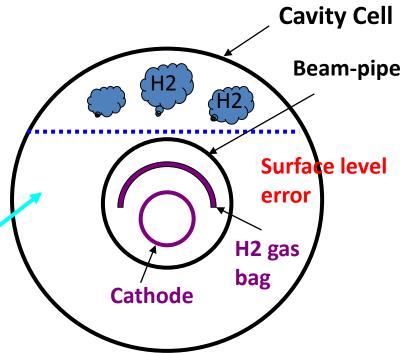
Cathode and H2 Gas Bag

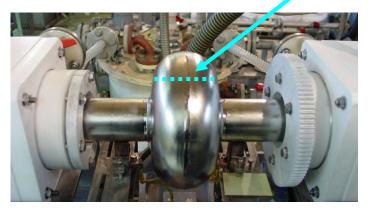


Level control of EP electrolyte

EP acid output valves for level control





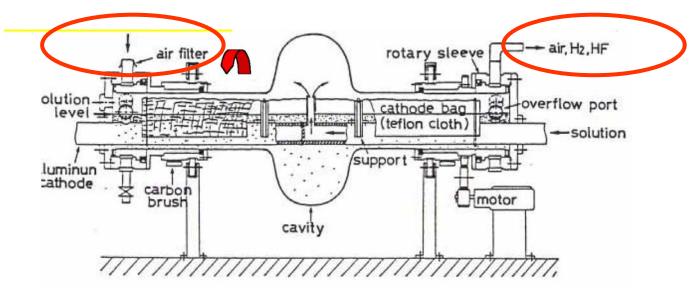


Failure of surface level of EP acid causes accumulation of H₂ gas in the cell. Q-disease

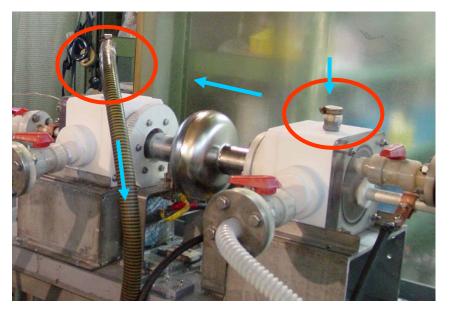
(Low Q from low gradient)

Surface level error

Exhaust/Suction of Hydrogen Gas



Exhaust/Suction of H₂ gas

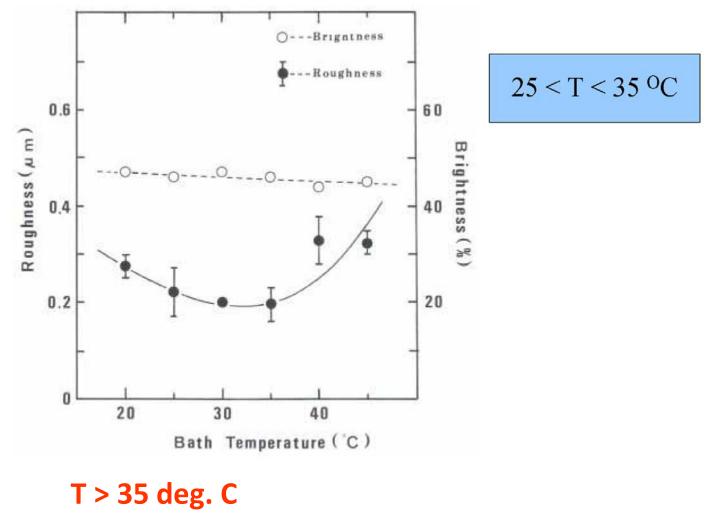


Evaporation of HF from EP electrolyte

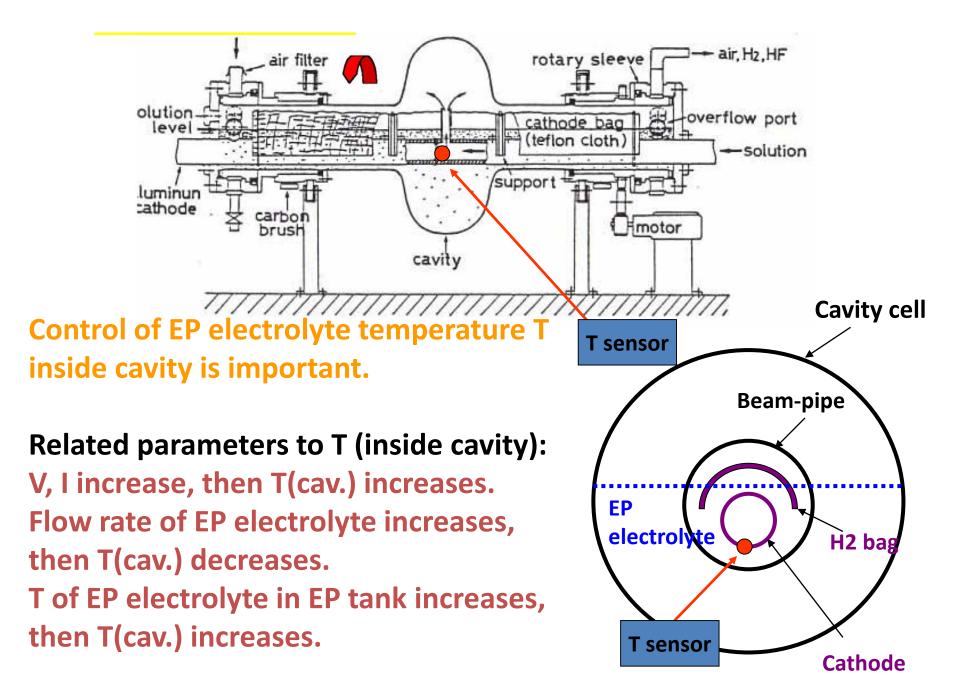
This valve is open during the EP process in order to exhaust hydrogen gas. But after EP process finished, this valve should be closed. If you keep this valve open, the HF evaporates and you lose HF concentration in EP electrolyte.



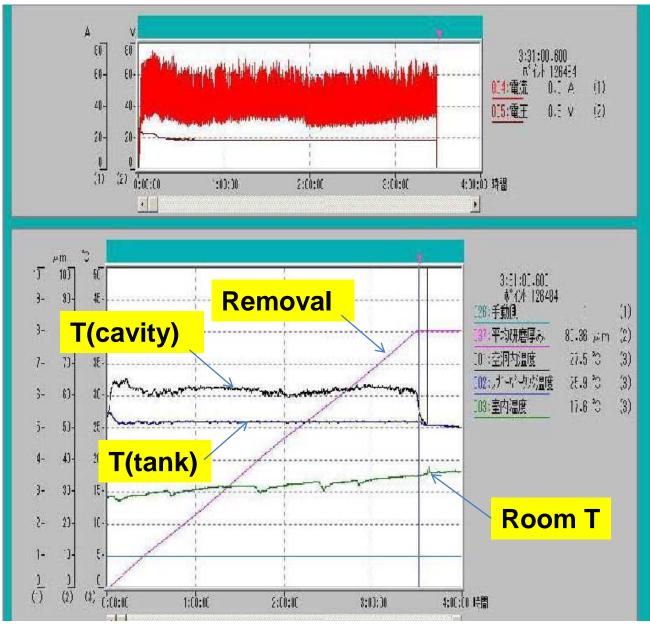
Temperature Effect on EP Finishing



=> Higher risk for Q-disease



Parameters (EP)



Electro-Chemical polishing inside 9-cell cavity

Electro-Chemical Polish Use Sulfuric acid + HF mixture Apply voltage between center AI electrode and Nb cavity

Optimize parameter for smooth surface without sulfur residual particle voltage and temperature are key parameter

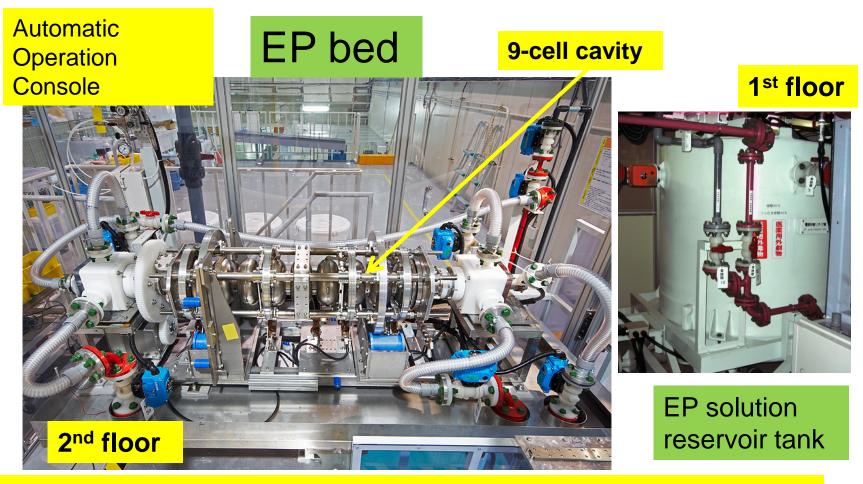
Successive rinsing is another key technology

©Rey.Hori

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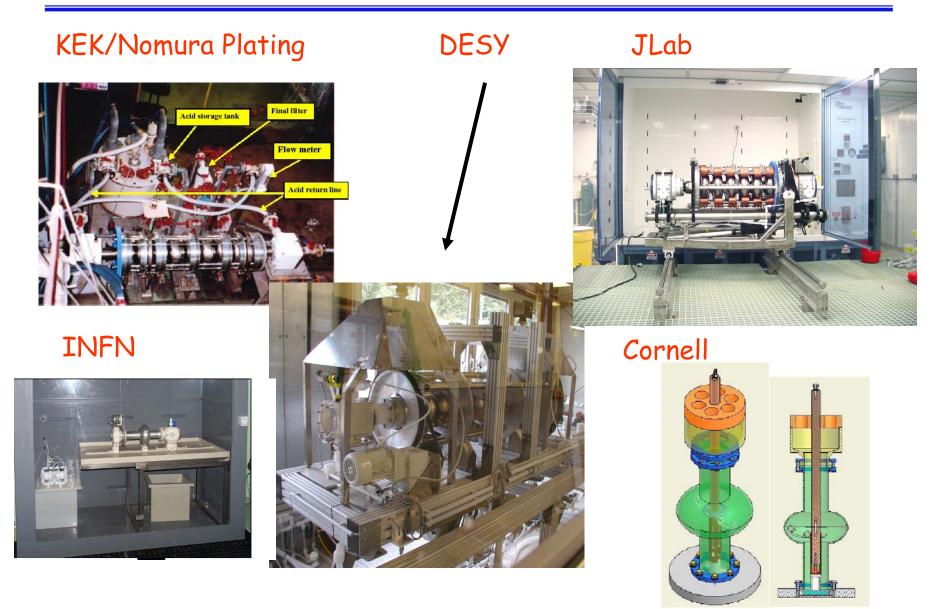
Electro-polishing facility at STF/KEK





EP facility at KEK EP acid: HF + H₂SO₄, Aluminum anode, surface removal speed: 20µm/hour, V ~18V, I ~270A, T ~30degC (for 9-cell), cavity rotation: 1 rpm.

Various EP systems in the world



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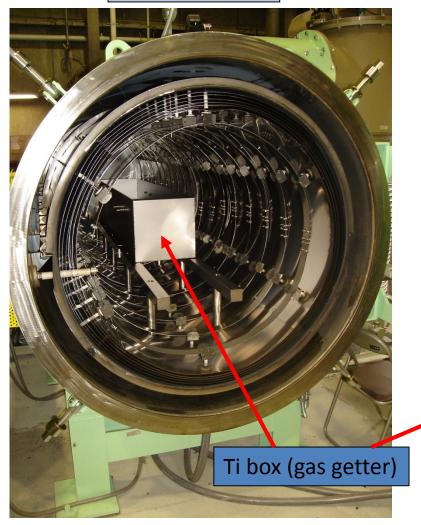
Annealing / Degassing



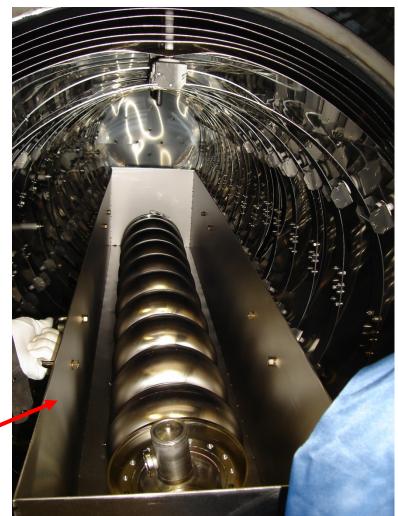
Annealing / degassing furnace at KEK: Two 9-cell cavity can be processed at once. Designed to consider about the supper-structure (Super-structure is consisting of two connected 9-cell cavities with one input-coupler).

Annealing / Degassing



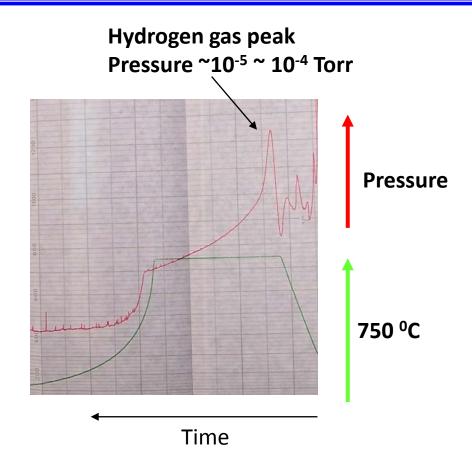


Cavity is set in a Ti box (gas getter)



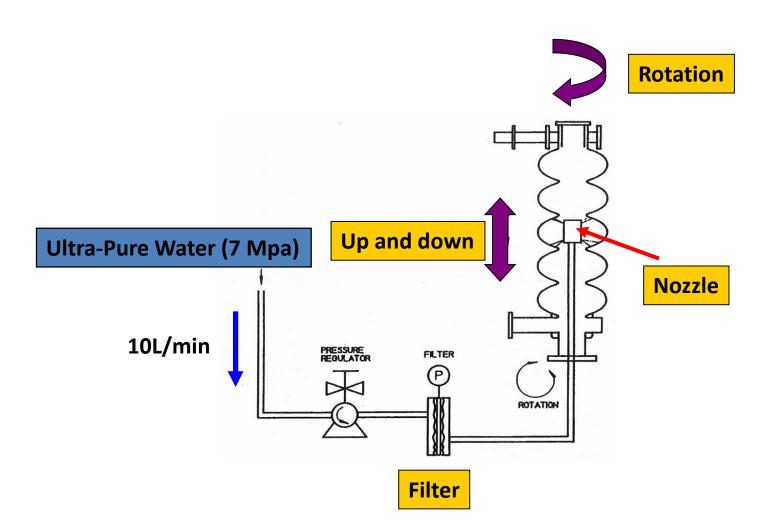


Annealing / Degassing



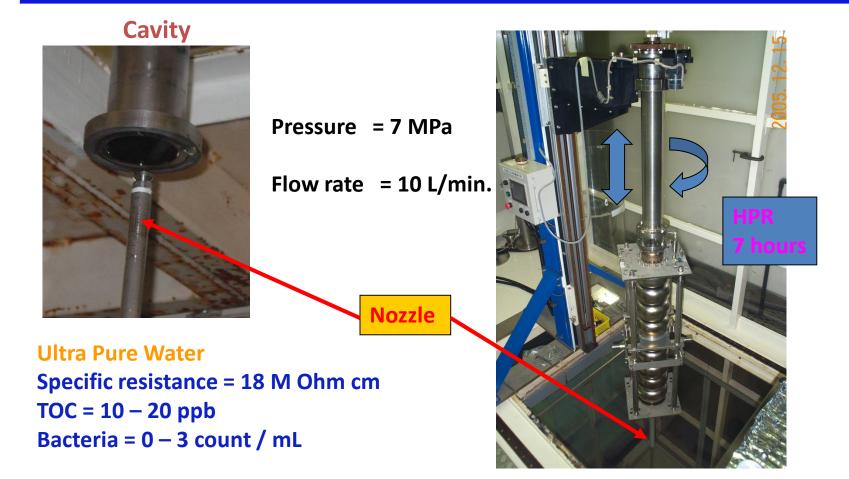
KEK recipe : 750 °C, 3 hours. Hydrogen gas can be degassed. Hydrogen in the Nb material causes Q-disease that degrades the Q value.

High Pressure Rinse (HPR)



High Pressure Rinse (HPR)





HPR is a strong tool to clean up the inside of cavity.

Various High Pressure Rinse (HPR) machines



DESY-System



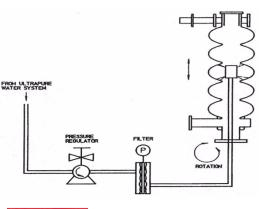


Jlab HPR Cabinet





KEK-System

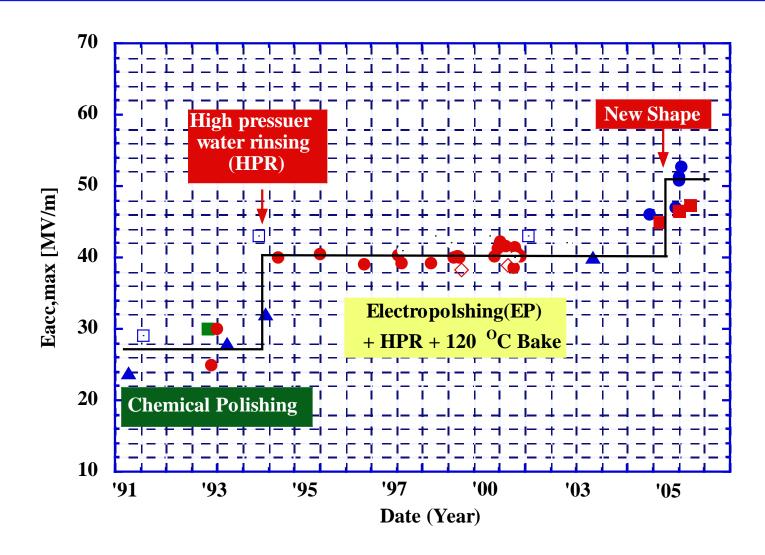






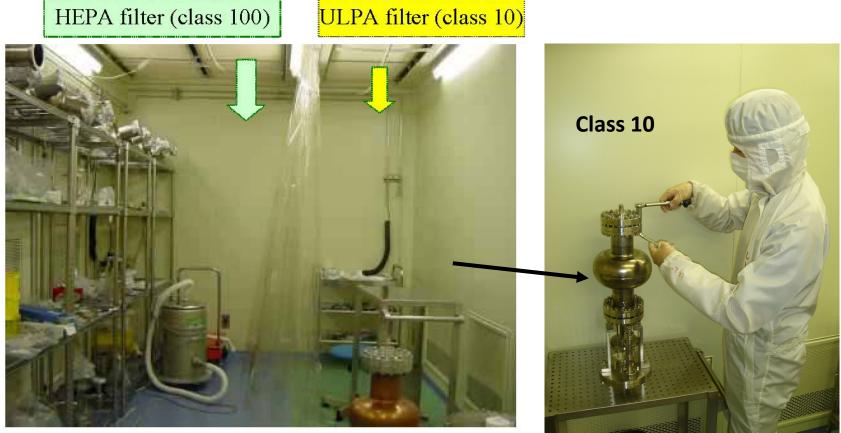


Breakthrough by HPR



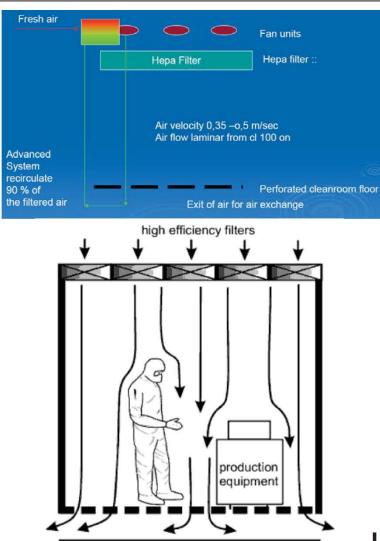
Assembly in Clean Room

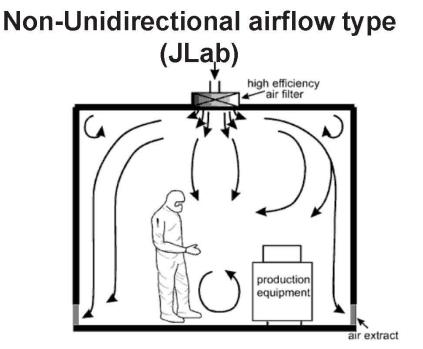
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Clean-room

Type of Cleanrooms





Unidirectional airflow type (DESY)

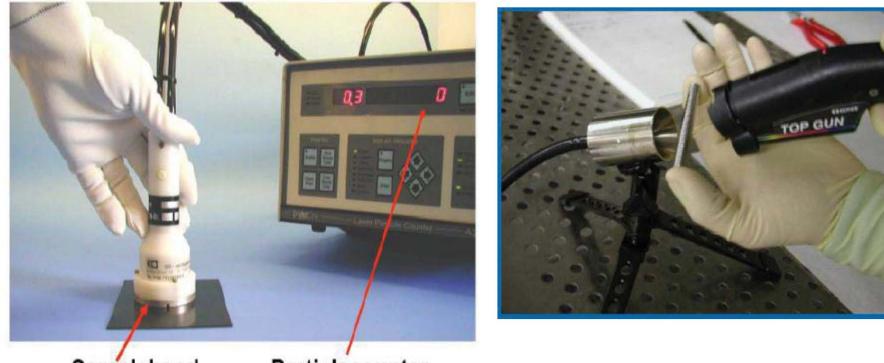


Cleanroom Classification

ISO Classification number	Maximum concentration limits (particles/m ³ of air) for particles equal to and larger than the considered sizes shown below										
	>=0.1µn	>=0.1µm>=0.2µm		>=0.3µm		>=0.5µm		>=1µm		>=5.0µm	
ISO Class 1	10	2									
ISO Class 2	100	24	10		4						
ISO Class 3	1 000	237	102		35		8				
ISO Class 4	10 000	2 370	1 020		352		83				
ISO Class 5	100 000	23 700	10 20	00	3 520		832		29		
ISO Class 6	1 000 000	237 000	102 (000	35 200		8 320		293		
ISO Class 7					352	000	83 20	00	2 930)	
ISO Class 8					3 52	0 000 0	832 (000	29 30	00	
ISO Class 9					35 2	35 200 000		8 320 000		293 000	
ISO 14644-1 Classes FS 209 Classes	Class 3 Class1	Class 4 Class 10 Cavity assembly		Class 5 Class 100 Cleanroo for SRF		Class 6 Class 10 n		Class 7 0 Class 10, 000		Class 8 Class 100, 000	

Particle Counters

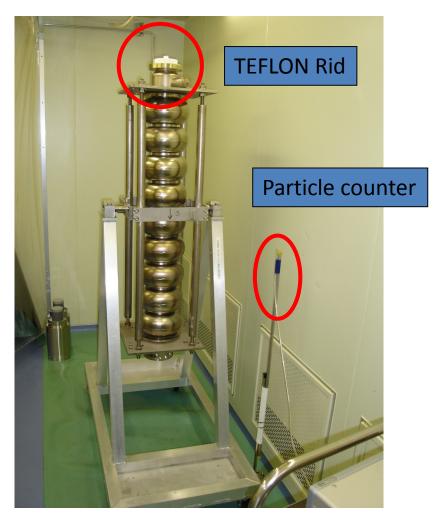
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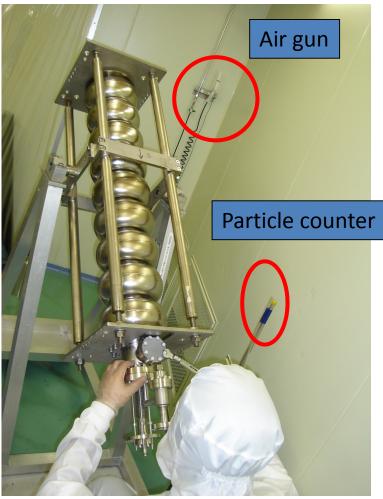


Samplehead

Particlecounter

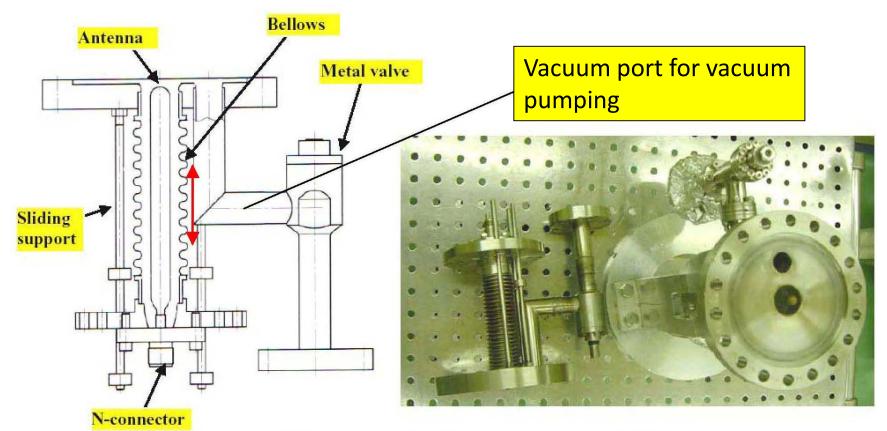
Assembly in Clean Room





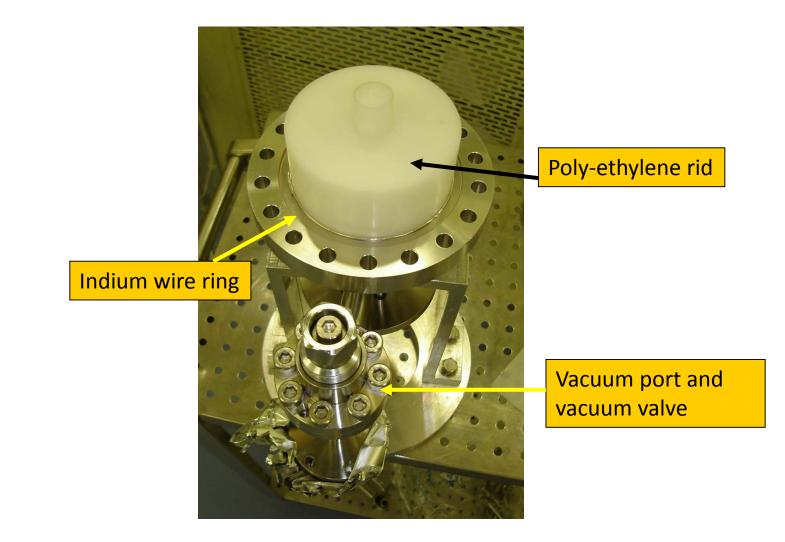


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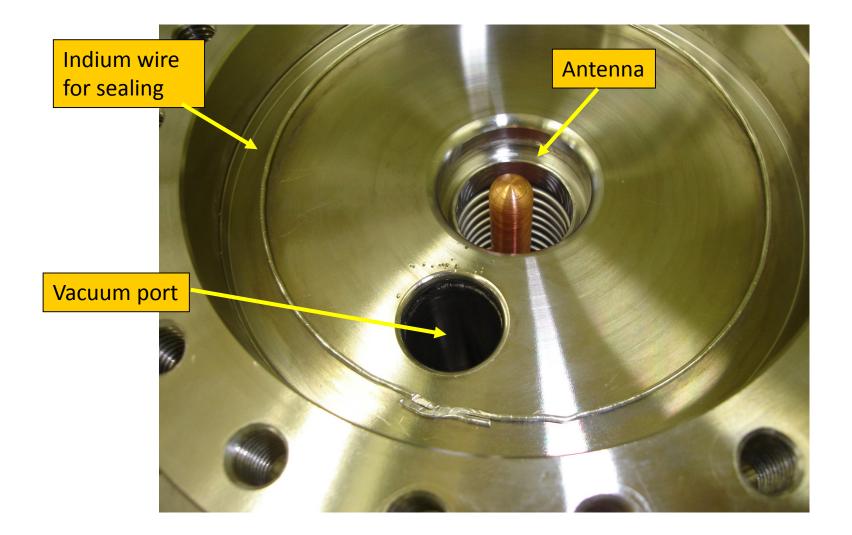


Variable input coupler for the vertical test in KEK

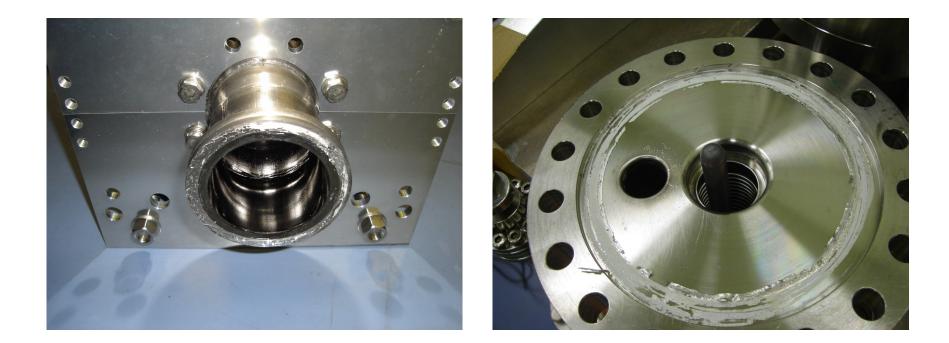
Indium Sealing



Indium Sealing



Indium Sealing



Indium is soft material. So the operation is complicated. Even when disassembly of cavity, the removal of indium is not easy work. And because the melting point of indium is low : T = 156 C, you should pay attention to the temperature control of In-situ baking (T=120 – 140).

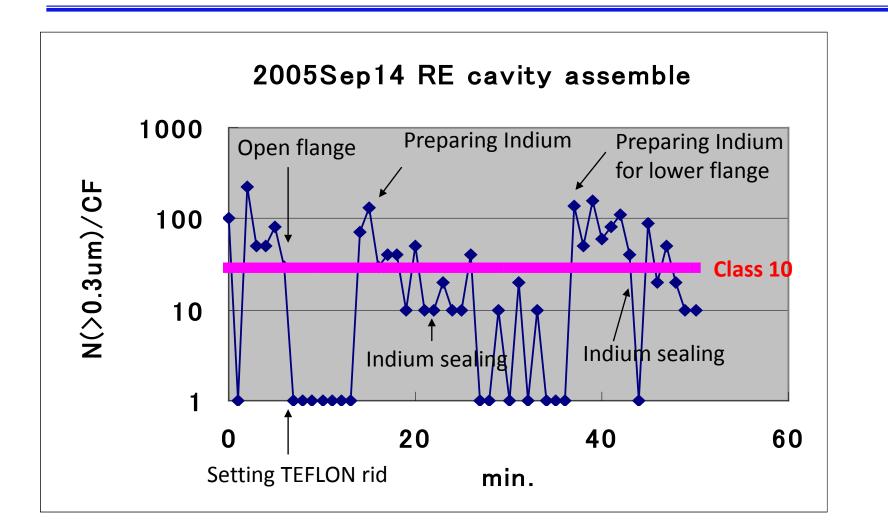
Assembly in Clean Room





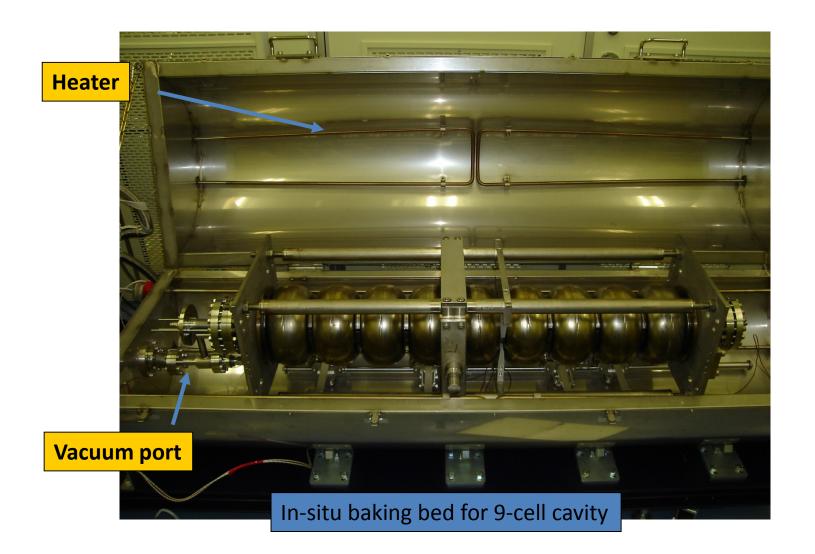


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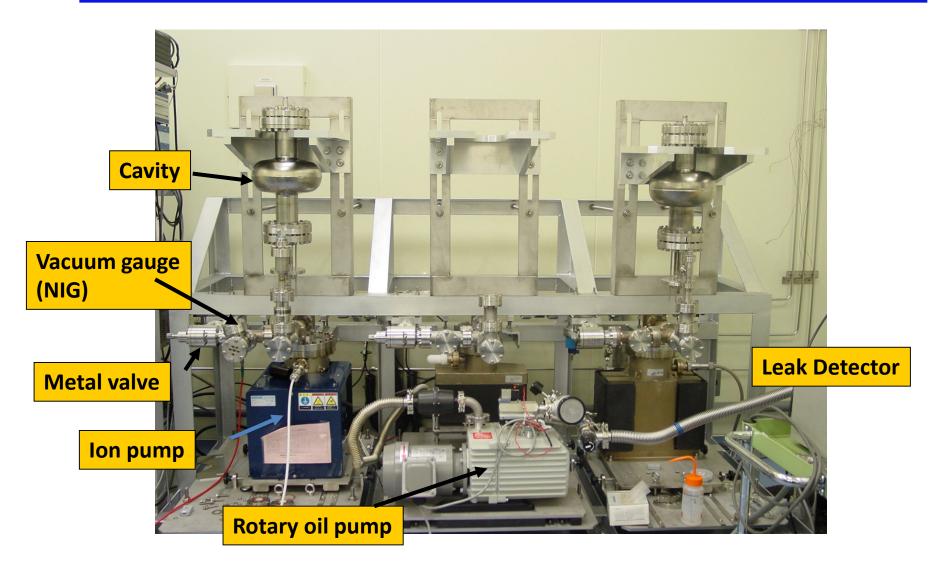
28 particles (>0.3um) / CF = class 10

In-situ Baking System for 9-cell Cavity



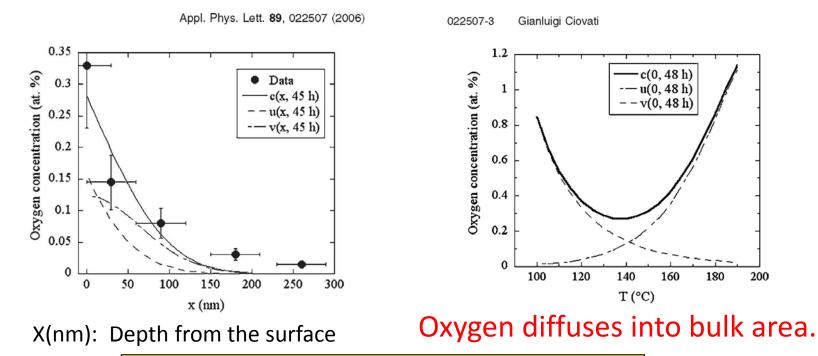
Vacuum System for In-situ Baking





Diffusion of Oxygen by In-situ Baking

The SC current inside the cavity has the depth of ~40 nm (London Penetration Depth)



c = total O concentration (~ u + v) u : O concentration from Nb₂O₅ decomposition (Nb₂O₅ => NbO₂ =>NbO) v : Initial O concentration after diffusion by baking

A hypothesis is that oxygen causes Q-drop only at high gradient.